Minority bulk and surface proton conduction in ceramic positrodes for proton ceramic electrochemical cells

Introduction

The positrode is critical for proton ceramic electrochemical cells for **hydrogen** and ammonia, as a major contribution to the over-potentials and hence losses in the whole cell. It is challenging to characterise the proton transport in predominantly electronic conductors.

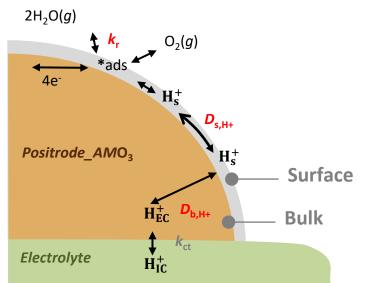
We want to establish theory and methodology for **measuring minority protonic** conductivities in electronic conductors. The results will be used as input to other project which perform computer simulations to seek strategies for optimization and effects on electrodes in scaled-up cells.

Primary objective

• Proton conduction in the bulk and on surfaces of positrode materials.

Secondary objectives

- Surface protonic conductance by proton flux.
- Proton conduction on conductive metal oxides.
- Brick layer model for mixed electronic and protonic conductive positrodes.



$$j_{\rm H^+} = -D_{\rm H_b^+} \frac{dc_{\rm H_b^+}}{dx} = -D_{\rm H_s^+} \frac{d\gamma_{\rm H_s^+}}{dx}$$

$$G_{\mathrm{H}_{\mathrm{S}}^{+}} = F \gamma_{\mathrm{H}^{+}} u_{\mathrm{H}^{+}} = F \gamma_{\mathrm{H}^{+}} \frac{F D_{\mathrm{H}^{+}}}{RT}$$



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Related projects: FME HYDROGENi

I am doing PhD in Materials Chemistry at the Electrochemistry group with Truls Norby as my supervisor.

Bachelor in Environmental Engineering -Huazhong University of Science and Technology, China.

Master of Research in Green Chemistry: Energy and Environment - Imperial College London, UK.

Estimated progress of the PhD project:

Publications

Wang, L.; Gu, Y.; Wei, J.; Wu, X. Li-Ni-Co-Mn oxides powders recycled from spent lithium-ion batteries for OER electrodes in CO2 reduction. *Environ. Technol.* Innov. 2020, 18, 13, Article. *DOI:* 10.1016/j.eti.2020.100732.

Dissertation of MRes

Using DFT to calculate the dual-atom catalysts for oxygen reduction reaction.



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